

SECTION A

1. 40 W bulb has higher electrical resistance.

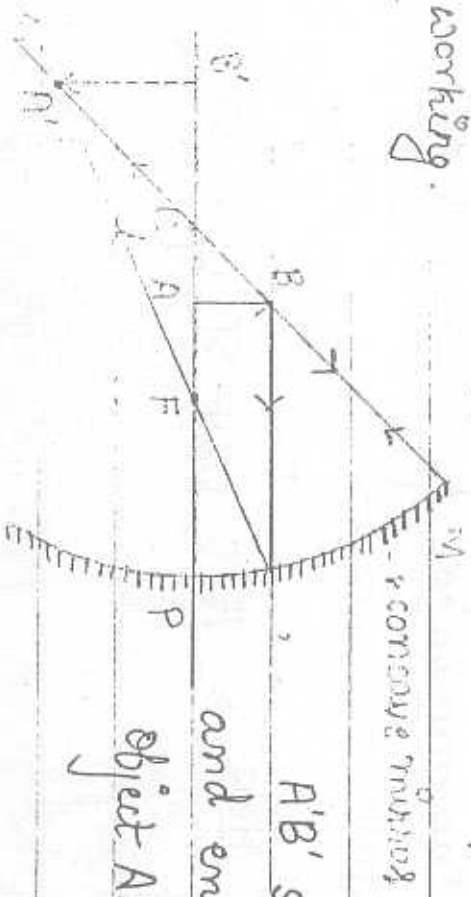
We know,  $R = \frac{V^2}{P}$   $\therefore R \propto \frac{1}{P}$ , Hence bulb having less power has higher electrical resistance.

2. Series arrangement is not used for connecting domestic appliances because :-

(1) In series, same current flows in circuit but different appliances like fan & heater require different currents.

(2) If one component fails, other components (appliances) also stop working.

3.



A'B' shows real, inverted and enlarged image of object AB

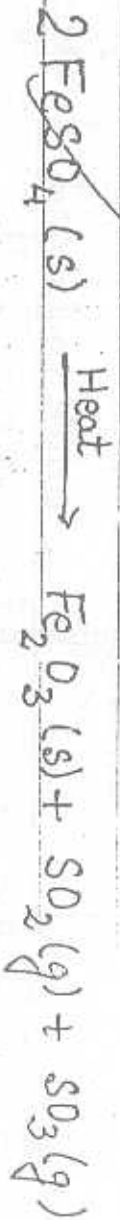
4 Flow of acid rain water into a river makes the survival of aquatic life in the river difficult because-

(A) Acid rain water has pH below 5.6. Hence it reduces the pH of river water.

(2) Aquatic organisms survive in pH range of 7.0-7.8. Hence their survival is reduced in reduced pH becomes difficult.

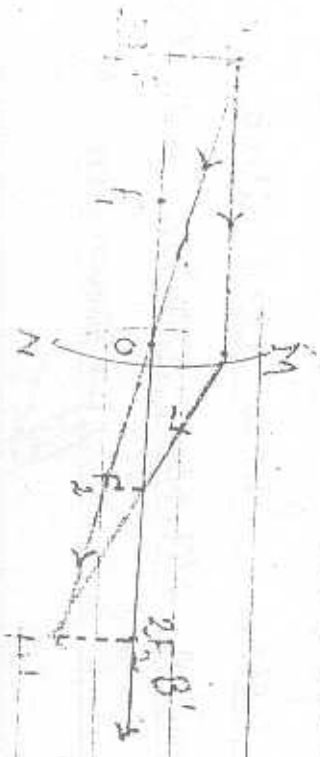
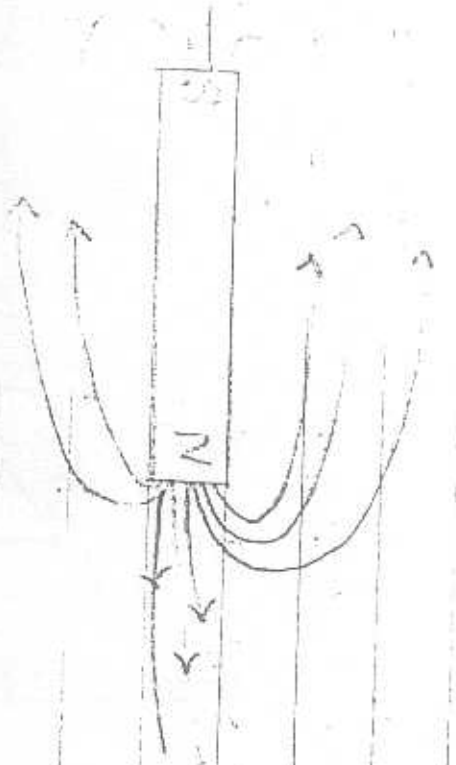


Energy required in our body is provided by Respiration. Since Respiration is accompanied by release of energy (heat), it is considered an exothermic reaction.



Properties of Magnetic Field lines -

- (i) They have both direction and magnitude. They originate from North to South.
- (ii) Magnetic field lines are closed curves.
- (iii) Magnitude of magnetic field lines is more near poles (closeness of magnetic poles).



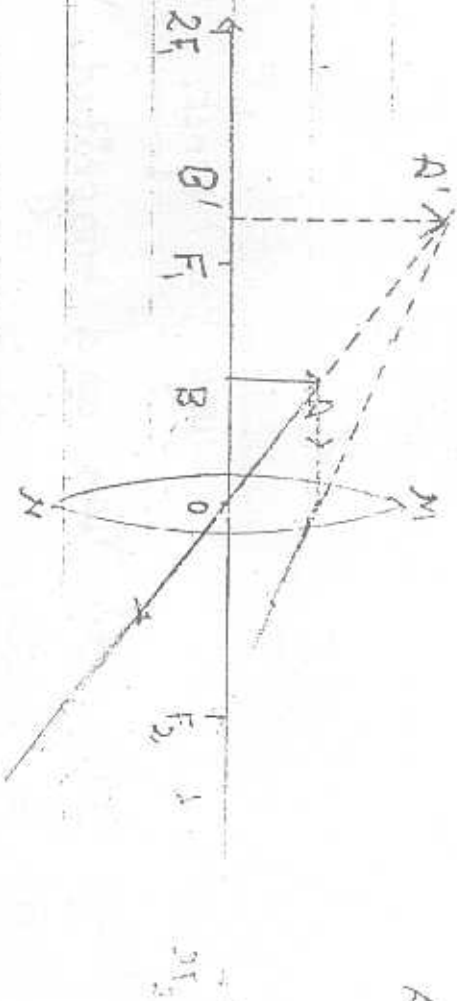
AB - object

A'B' - image

Nature - real and inverted

Position - at  $2F_2$

Size - Same as object



(2) between  $F_1$  and  $O$ .

AB - Object

$A'B'$  - Image

Nature - Virtual and Erect

Size - Enlarged

Position - Same side of lens

9

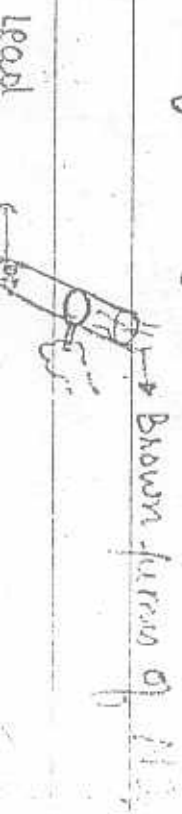
Example of Decomposition Reaction -



Brown fumes

Activity

- (1) Take 2 spatula of Lead Nitrate in a test tube
- (2) Using a holder heat Lead nitrate on a Bunsen burner
- (3) Brown fumes of Nitrogen dioxide are observed.



Chemical formula of bleaching powder :  $CaOCl_2$

[Calcium oxychloride]

Bleaching powder is prepared by passing  $Cl_2$  on dry slaked lime,

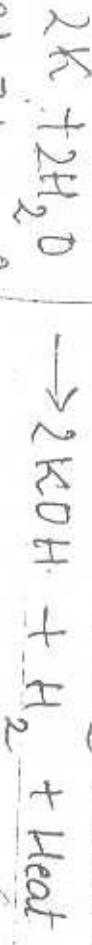


It is used for bleaching wood pulp in paper industries,

11. • Two metals which react slowly with cold water : Sodium (Na), Potassium (K)

• Observations:-

(1) Following reactions take place when they are dropped into water:-



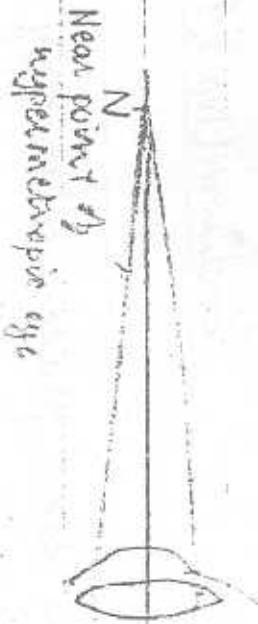
(2) There is strong fizzing sound of Hydrogen

(3) Evolved Hydrogen gas immediately catches fire.

(4) Beakers become hot since the reaction is exothermic

(b) Hypermetropia - (Far Sightedness) - The object of eye can not see nearby objects clearly. The near point of such an eye is more than 25 cm.

(1) Hypermetropic Eye



(2) Correction with convex lens

(a) Overloading

(i) When too many devices are connected in a single circuit (socket), the circuit gets damaged. This is known as overloading.

(ii) It happens when many appliances are connected in same socket or there is some hike in power supply.

Short-circuiting

(i) When live and neutral wires come in direct contact, the amount of

current increases, this <sup>heat</sup> is called short circuiting damaging the circuit.

(ii) It happens when insulation cover of wires are damaged or <sup>wires</sup> the appliances is not working properly.

(b) Toasters are made of an alloy rather than pure metal because:-

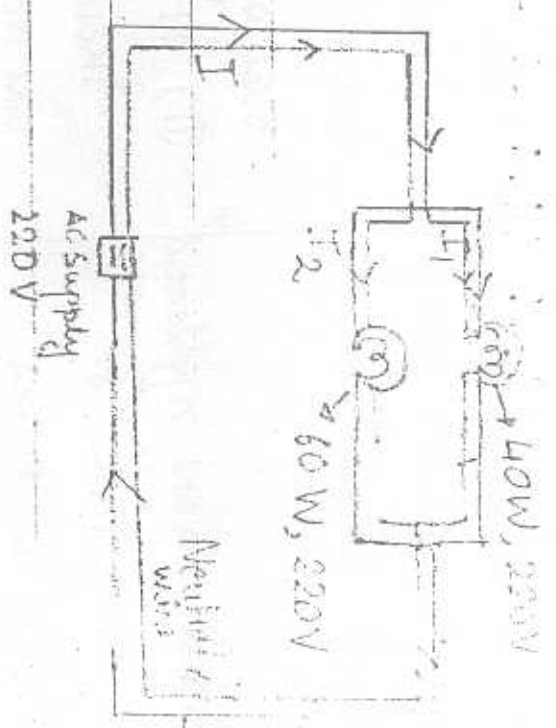
- (1) Alloy generally have higher resistivity than metals. Thus they can convert more electricity into heat, energy.
- (2) Alloy don't oxidise (burn) even at high temperatures.

(2) Thick, on, ohmic, low resistance than metals.



14.

(a)



(b) Current drawn by 40 W bulb =  $I_1 = \frac{P}{V}$  [  $P = VI$  ]

$$\therefore I_1 = \frac{40}{220} \text{ A} = \underline{\underline{0.18 \text{ A}}}$$

Current drawn by 60 W bulb =  $I_2 = \frac{P}{V} = \frac{60}{220} \text{ A} = \underline{\underline{0.27 \text{ A}}}$

$\therefore$  Total current drawn from circuit,  $I = I_1 + I_2 = (0.18 + 0.27) \text{ A} = \underline{\underline{0.45 \text{ A}}}$

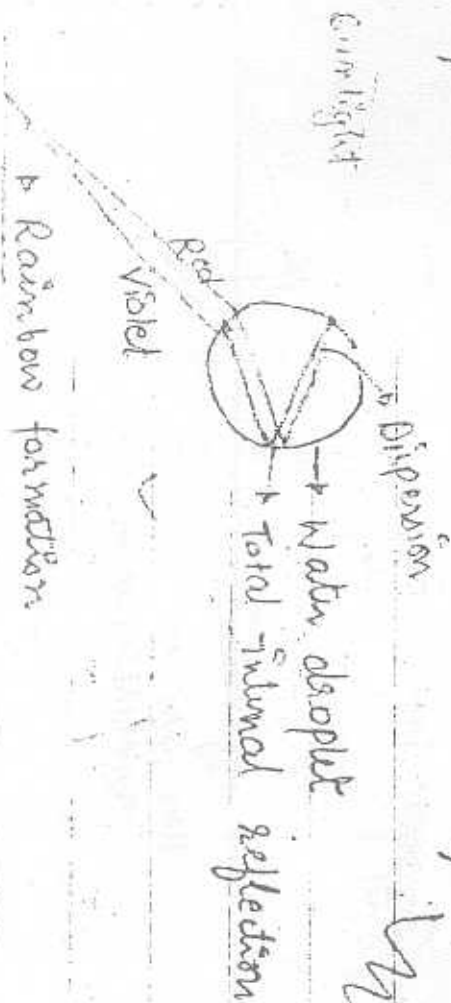
(c) Energy consumed by 40 W bulb in 1 hour =  $P \times T = \underline{\underline{40 \text{ Wh}}}$



∴ Total energy consumed = 40 kWh + 60 kWh  
= 100 kWh

$$= \underline{\underline{0.1 \text{ kWh}}} \quad [∵ 1 \text{ kWh} = 1000 \text{ Wh}]$$

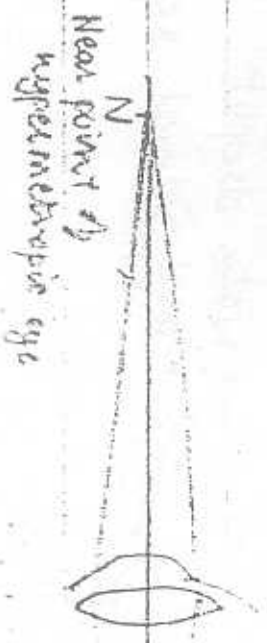
15. (a) Dispersion - When a beam of light passes through a glass prism, it splits into seven colours (VIBGYOR). This phenomenon is known as dispersion of white light.



Tiny water droplets in the sky act as small prisms. When sunlight passes through them, they first refract, then disperse, reflect it internally and finally refract it again to form rainbow.

(b) Hypermetropia - (Far sightedness) - The defect of eye in which a person can see far objects clearly but cannot see nearby objects clearly. The near point of such an eye is more than 25 cm.

(1) Hypermetropic Eye



(2) Correction with convex lens

16. (a) Element in

(i) I<sup>st</sup> group, II<sup>nd</sup> period - Sodium (Na)  
(ii) III<sup>th</sup> group, III<sup>rd</sup> period - Fluorine (F)

(b) (i) Oxide of Nitrogen -  $N_2O_5$   
[General -  $R_2O_5$ ]

(ii) hydride of Oxygen -  $H_2O$   
[General -  $RH_2$ ]

(c) 5<sup>th</sup> group VIII of periodic table, cobalt with atomic mass 58.9 appears not before Nickel (58.7) because cobalt is chemically similar to Rhodium and Iridium and Nickel is chemically similar to Palladium and Platinum.

- (d) Beside Gallium two other elements discovered are -  
(1) Germanium [Eka-Silicon]  
(2) Scandium [Eka-Boron]

(e)

Atomic mass of Li = 6.9	Atomic mass of Na = 23
Atomic mass of K = 39.1	

∴ Average Atomic mass of Li & K =  $\frac{6.9 + 39.1}{2} = 23$

∴ Average Atomic mass of Li and K = ~~Average~~ atomic mass of Na

Conclusion-

- (1) Dobereiner's Triads occur in Mendeleev's Periodic Table also.
- (2) When chemically similar elements are arranged in increasing order of atomic mass, atomic mass of middle element is the average of atomic mass of other two elements. [Dobereiner's Law of Triads]

SECTION B

There are following effects of DNA copying not being accurate

not exactly similar to parent.  
(2) When DNA changes are very drastic, the new cell is  
(3) It leads to survival advantage in some cases

18. Obtaining of  $CO_2$  :-

(1) Autotrophs obtain  $CO_2$  from air through their stomata [ openings ]. present in leaves, stem, Obtaining of  $N_2$  :-  
(1) Autotrophs obtain  $N_2$  dissolved in water from soil through their roots.  $N_2$  in the form of Nitrate and

19. Reasons for replacing firewood by alternate sources of energy :-

- (1) It produces much smoke on burning of them etc. causes air pollution
- (2) It releases very small amount of energy on burning of low calorific value.

- (3) It leaves a residue [wood ash] on burning.
- (4) Using firewood as fuel causes deforestation.

20. (a) Winds - To extract energy from winds, large wind mills have to be erected. But these mills can only be established in those areas where wind flows for larger part of year. Minimum speed - 15 km/h. It also requires high level of maintenance.

#### Tidal

(b) Tides - (1) Energy can be extracted only in few places.

(2) Building <sup>structures</sup> in sea-water requires high cost.

21. Characteristics of Biogas :-

- (1) It contains 75% methane and thus gives a large amount of heat on burning. [High calorific value],
- (2) It is comparatively smokeless.

22. Two sexually transmitted diseases -

HIV-AIDS, Syphilis  
Gonorrhoea

Prevention :-

- (1) Use of condom for penis like condom while having sex.
- (2) Avoid sex with unknown people.

23. Fossils are the prehistoric remains of dead plants and animals trapped in rocks underneath.

Information that fossils provide us on evolution -  
(1) It helps to study organ structure of pre-historic animals.

(2) It tells us which type of animals lived at what time on Earth.



24.

(a) Blood vessels - (Arteries, Veins, Capillaries). They carry from heart to body organs and also from body organs to heart.

(b) Blood platelets - They help in plugging any leak blood vessels (Blood coagulation)

(c) Lymph - (Tissue Fluid) - It carries digested fat and proteins from intestine to veins. It drains extra-cell fluid back to blood.

(d) Heart - Heart is a pumping organ which helps to pump blood to our body organs.

25.

Aerobic respiration

1. It takes place when glucose is completely broken down in presence of oxygen.

Anaerobic respiration

1. It takes place when glucose is partially broken down in absence of oxygen.

[in mitochondria]  
3- It takes place in our body cells to give  $\text{CO}_2$  and  $\text{H}_2\text{O}$ .

3. It takes place in yeast to give Ethanol and  $\text{CO}_2$  and in our muscle cells to give Lactic Acid.

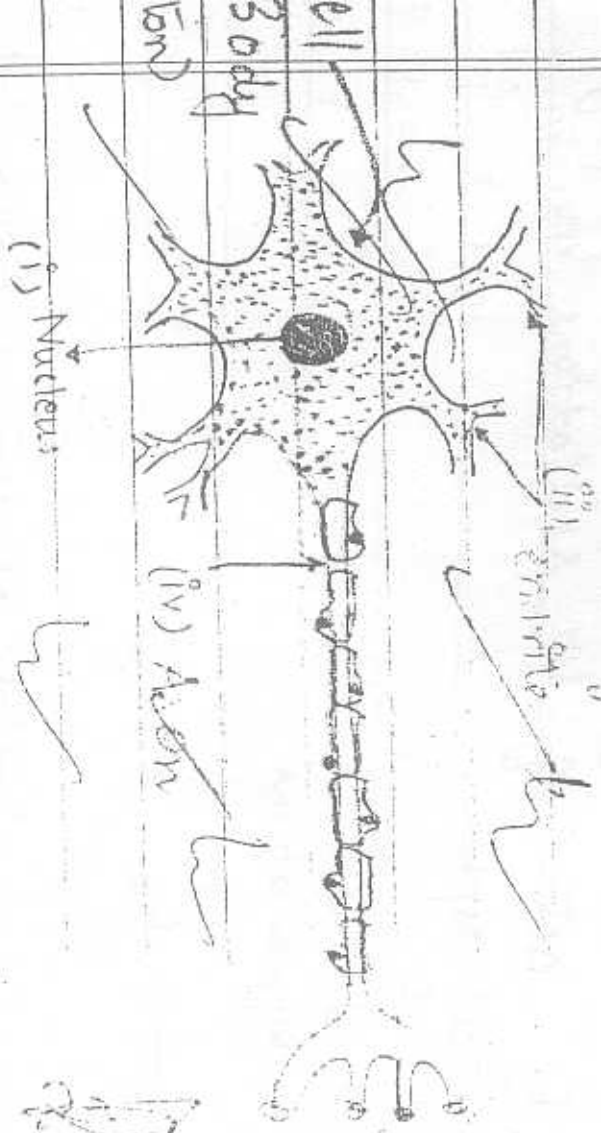
6. Ozone ( $\text{O}_3$ ) is formed by action of UV radiations on Oxygen in upper atmosphere.



Ozone is a protective layer which prevents us from harmful Ultra-Violet radiations. UV radiations cause skin cancer in humans. Hence damage to ozone is a cause of concern for us.

• Ozone ( $O_3$ ) layer is damaged by Chloro-Fluoro Carbon (CFC) which are used as refrigerants & fire extinguishers.

Q1. (a) Structure of Neuron



(i) Part of Neuron where information is acquired.

Dendrite Tip